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Chemical Reactions The standard representation of a chemical reaction shows an arrow pointing from the

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reactants to the products: For most chemical reactions in this book, solids are labeled (s), liquids (l), and gases (g). The numerical coefficients in front of the chemical formulas express the moles of each compound or element.

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Chemical reaction that releases energy=  
Example= jellyfish, glowsticks,  
respiration Temperature= increasing.  
Respiration. Exothermic process by  
which living things release energy from  
glucose and oxygen and produce carbon  
dioxide and water. Photosynthesis  
equation. Products=Oxygen and sugar  
Reactants= Carbon dioxide, water, and

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energy Photosynthesis= Endothermic reaction.

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Section 4.1 Chemical Reactions and Chemical Equations Goals To describe the nature of chemical reactions. To show how chemical reactions can be

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described with chemical equations. To show how to balance chemical equations. This section starts with a brief description of chemical reactions. The introduction to chemical

## **Chapter 4 An Introduction to Chemical Reactions**

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guide. STUDY. PLAY. Terms in this set  
(...) chemical formula. used to represent  
the composition of elements and  
compounds (pure substances)  
qualitative. information of a formula  
shows the types of atoms (or ions) that  
make of the the substance. quantitative.

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Name CHAPTER Date Class STUDY  
GUIDE Section 9.2 Classifying Chemical  
Reactions In your textbook, read about  
synthesis, combustion, decomposition,  
and replacement reactions. Assume that  
Q, T, X, and Z are symbols for elements.  
Match each equation in Column A with  
the reaction type it represents in Column

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B. Column A 2.

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## **CHEMISTRY: Study Guide "Chemical Reactions and Equations ...**

Chemistry Chemical Reactions Study  
Guide. Reactant. Product. Catalyst.  
Coefficient. Substance(s) present at the  
beginning of a chemical reaction;....  
Substance(s) formed by a chemical  
reaction; on the right side.... A

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substance that increases the rate of a chemical reaction but....

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Chemical Reactions & Equations: Just as atoms and ions combine in very specific ways, molecules and compounds react with each other in definite quantities.

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Learn how to tell whether or not a reaction can occur and what the products of a reaction will be. Write balanced chemical equations to describe reactions.

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STUDENT STUDY GUIDE FOR 8TH GRADE CHEMISTRY. Welcome to chemistry.

Chemistry is the science of matter and its interactions. Everything around you is made of atoms – atoms and their chemical combinations, molecules.

Everything you can see, touch, smell, and taste is made up of chemicals. So,

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the science of chemistry is a very wide ranging science.

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In the chemical reaction of carbon and oxygen to form carbon dioxide, the elements are in a definite 1:2 ratio—one atom of carbon with two atoms of

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oxygen forming the compound carbon dioxide. When several atoms are so tightly bonded together that they physically behave as a unit, the unit is called a molecule. (See Figure 1.)

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Chapter 4: An Introduction to Chemical

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Enduring Understanding 3.B: Chemical reactions can be classified by considering what the reactants are, what the products are, or how they change from one into the other. Classes of chemical reactions include synthesis, decomposition, acid-base, and oxidation-reduction reactions.

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Atoms of different elements combine in simple whole number ratios to form chemical compounds. In chemical reactions atoms are combined, separated or rearranged to form new compounds. Dmitri...



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Chemistry: Chemical Reactions Study Guide. by ksaulsgi, Mar. 2015. Subjects: chemistry, chemistry i, chemical reactions . Click to Rate "Hated It" ... an expression representing a chemical reaction; the formulas of the reactants

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(on the left) are connected by an arrow with the formulas for the products (on the right)

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